Name	Slayter Box	
Examination III Retake		December 7, 2012

Intermediate Organic Chemistry (CHEM 251-03) Dr. Fantini

OPTIONAL RETAKE EXAM 3

Please do not open until instructed

You have two hours to complete this examination.

Intermediate Organic Chemistry (CHEM 251-03) Dr. Fantini

Examination III December 7, 2012

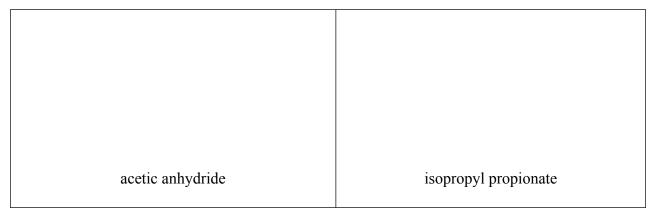
OPTIONAL RETAKE EXAM 3

Notes:

- This exam consists of **5 questions**. Please check to make sure that you have a complete copy of the exam.
- Please do not simply give me answers. Give me well-supported answers. Answers that are not backed by explanations will receive minimal credit.
- Please write clearly; if I can't read your answer, I can't give you credit for your answer.
- Please note that different questions are worth different numbers of points. Plan your time accordingly.
- Remember to include units and significant figures where appropriate.
- No books or notes are to be used on this exam.
- Please do NOT share calculators; if you need a calculator but do not have one, please let me know!
- If you feel that you would be better able to answer **any** question if you had additional information, please do not hesitate to ask for it. I will happily provide any information that I feel will help you answer the question without compromising the efficacy and fairness of the test.

Question	Possible	Score
1	6	
2	10	
3	40	
4	20	
5	24	
TOTAL	100	
	Approx. Letter:	

61. Nomenclature. Please give name for structure or structure for name.

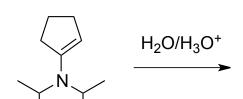


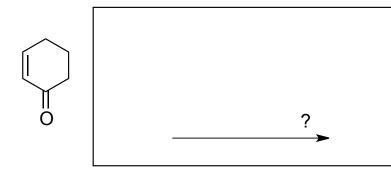
102. Complete these reactions (the example and the first three are proton transfers) and *CIRCLE* the side (products or reactants) that are favored at equilibrium.

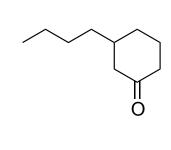
Fill in any of the missing starting material(s), reagent(s), and/or dominant *organic* product(s) for each single reaction. Please specifically denote all stereochemistry.

O NaOH, H₂O

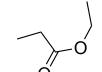
 $O H_2O/H_3O^+$





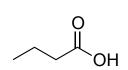


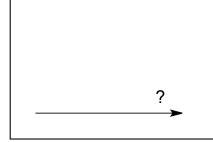
Question 3 continued.

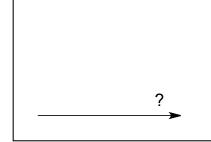


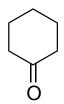
- 1. 2 CH₃Li 2. H₂O/H₃O⁺

- $1.\; LiAlH_{4}$
 - 2. H₂O/H₃O⁺









- 1. LiNHiPr₂ ("LDA") 2. (CH₃)₂CHCH₂CH₂CI

Question 3 continued.

$$N$$
-CH₃ 1. LiAlH₄ \longrightarrow 2. H₂O/H₃O⁺

Question 3 continued.

- 1. LDA O 2. H O slowly 3. H₂O/H₃O⁺
 - Ag_2O, NH_3
 - PCC
 CH₂Cl₂
 OH

 CC = pyridinjum chlorochromate

PCC = pyridinium chlorochromate

- 204. Short syntheses.
- (a) Propose a synthesis of this compound from *cyclopentanone*.

(b) Propose a synthesis of this compound from *cyclopentanone*.

(c) Synthesize the product shown at right by the *acetoacetic ester synthesis*.

(d) Use a *protecting-group strategy* to accomplish this synthesis:

$$HO \longrightarrow HO \longrightarrow OH$$

245. Please draw a stepwise electron pushing mechanism for the reactions shown below.

Problem 5, continued.